

United States Sent and Trademark Office

UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address O AMMISSIS ORES FOR PATENTS FO USE 1499 Advandana Viggan 223/1/1450 www.nept.gov

APPLICATION NO	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO	CONFIRMATION NO
09.934,806	08.22.2001	Herman Uytterhoeven	212967	8829
23460	7590 08 13 2003			
LEYDIG VOIT & MAYER, LTD TWO PRUDENTIAL PLAZA, SUITE 4900 180 NORTH STETSON AVENUE			EXAMINER CHEA, THORL	
			1752	.7
			DATE MAILED: 08/13/2003	1

Please find below and/or attached an Office communication concerning this application or proceeding.

	Auglieght No	
	Application No.	Applicant(s)
Office Antique Comme	09/934,806	UYTTERHOEVEN ET AL
Office Action Summary	Examiner	Art Unit
	Thorl Chea	1752
The MAILING DATE of this communication a Period for Reply	ppears on the cover sheet v	vith the correspondence address
A SHORTENED STATUTORY PERIOD FOR REP THE MAILING DATE OF THIS COMMUNICATION - Extensions of time may be available under the provisions of 37 CFR in If the period for reply aspecified above, the maximum statutory are If the period for reply is aspecified above, the maximum statutory are Failure to reply, within the set or extended period for reply will, by statutory are reply aspecified above, the maximum statutory are Failure to reply, within the set or extended period for reply will, by statutory and the statutory are reply aspected by the statutory and the statutory are reply aspected by the statutory and the statutory are reply aspected by the statutory and the statutory are reply aspected by the statutory and the statutory are reply as the statutory and the statutory and the statutory are reply as the statutory and the statutory are statutory as the statutory are statutory as the statutory and the statutory are statutory and the statutory are statutory as the statutory and the statutory are statutory as the statutory and the statutory are statutory as the statutory are statutory as the statutory are statutory as the statutory and the statutory are statutory as the statutory are statutory as the statutory are statutory as the statutory and the statutory are statutory as the statutory as the statutory are statutory as the statutory as the statutory are statutory as the statutory as the statutory as the statut	I. 1 136(a) In no event, however may a sply within the statutory minimum of the d will apply and will expire SIX (6) MC title, cause the application to become.	reply be timely filed iffy (30) days will be considered timely NTHS from the maling date of this communication BANDONED (35 U S C § 133)
1) Responsive to communication(s) filed on 02	2 June 2003 .	
2a) This action is FINAL . 2b)	This action is non-final.	
3) Since this application is in condition for allocation accordance with the practice under the practice under the practice.		
Disposition of Claims		
4) Claim(s) 1-18 is/are pending in the application 4a) Of the above claim(s) is/are withdrawith a significant of the application of the		
	rawn nom consideration.	
5)⊠ Claim(s) <u>1-4 and 17</u> is/are allowed. 6)☐ Claim(s) 5.6,8-16 and 18 is/are rejected.		
7) Claim(s) is/are objected to.		
8) Claim(s) are subject to restriction and	Var alaction requirement	
Application Papers	or election requirement.	
9)☐ The specification is objected to by the Examin	ner.	
10) The drawing(s) filed on is/are: a) acc		the Examiner.
Applicant may not request that any objection to	the drawing(s) be held in abe	yance. See 37 CFR 1.85(a).
11) The proposed drawing correction filed on	is: a) approved b)	disapproved by the Examiner
If approved, corrected drawings are required in	reply to this Office action	
12) The oath or declaration is objected to by the B	Examiner.	
Priority under 35 U.S.C. §§ 119 and 120		
13) Acknowledgment is made of a claim for foreign	ign priority under 35 U.S.C	. § 119(a)-(d) or (f)_
a)⊠ All b)□ Some * c)□ None of:		
 Certified copies of the priority docume 	nts have been received.	
Certified copies of the priority docume	nts have been received in	Application No
 Copies of the certified copies of the pr application from the International I 	Bureau (PCT Rule 17.2(a))	
* See the attached detailed Office action for a li	st of the certified copies no	t received.
14) ☐ Acknowledgment is made of a claim for dome		*
a) The translation of the foreign language p 15) Acknowledgment is made of a claim for dome		
Attachment(s)		
Notice of References Cited (PTO-892) Notice of Draftsperson's Patent Drawing Review (PTO-948) Information Disclosure Statement(s) (PTO-1449) Paper No(s)	5) Notice	v Summary (PTO-413) Paper No(s) f Informal Patent Application (PTO-152)
The state of the s		

Art Unit: 1752

DETAILED ACTION

Claim Rejections - 35 USC § 102

 The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States

Claim Rejections - 35 USC § 103

- The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter perfains. Patentability shall not be negatived by the manner in which the invention was made.
- Claims 5-6, 14 and 16 are rejected under 35 U.S.C. 102(b) as anticipated by or, in the alternative, under 35 U.S.C. 103(a) as obvious over Gilliams et al (Gilliams).

Gilliam discloses an aqueous dispersion containing silver salt of aliphatic carboxylic acid and silver halide in column 17, example 3; column 17, example 6; in column 21, example 19; and column 4, lines 23-28 and in column 9, lines 21-40. Gilliam in column 15, Examples 1 discloses the use sodium hydroxide to provide the aqueous solution to pH of 8.7 and in column 17; Example 2 discloses the formation of silver halide in-situe using the conversion of silver behenate. The aqueous solution in Example 2, column 17 contains 0.079 moles and 0.022 mole of silver halide.

The aqueous solution claimed in the present claimed invention and that taught in Gilliams are identical, except that composition of the claimed invention contains ex-situ

Art Unit: 1752

silver halide whereas silver halide taught in Gilliam is made by in-situe process, but the

 $\ensuremath{\mathsf{pH}}$ of the composition are the same. Accordingly, it is asserted that the composition as

claimed is either anticipated or found obvious over Gilliam. Moreover, the invention as

claimed is related to the claiming of a material by a process. "(E)ven though product-

by-process claims are limited by and defined by the process, determination of

patentability is based on the product itself. The patentability of a product does not

depend on its method of production. If the product in the product-by-process claim is the

same or obvious from a product of prior art, the claim is unpatentable even though the

prior art product was made by different process." In re Thorpe 777 F.2d 695, 698, 227

USPQ 694, 966 (Fed. Cir. 1985).

4. Claims 8, 10, 11, 15 and 18 are rejected under 35 U.S.C. 102(b) as being

anticipated by Gilliams et al ('616). Note to Gilliam in column 17, Examples 2, and column 15, Example 1 which discloses the preparation of silver behenate and in-situe

silver halide by first forming an aqueous solution having an pH of 8.7 using sodium

hydroxide to control the pH thereof and then using conversion process by adding drop $% \left(1\right) =\left(1\right) \left(1\right) \left($

wise of potassium bromide. The aqueous solution contains 0.079 mole silver halide

behenate and 0,022 moles of silver halide. Gilliam taught the process and the aqueous

composition containing in-situe silver halide and light-insensitive silver salt of organic

acid as claimed and the claims invention lacks novelty.

5. Claims 9, 12-13 are rejected under 35 U.S.C. 103(a) as being unpatentable over

Gilliam et al ('616).

Page 3

Art Unit: 1752

Gilliam discloses to regulate the pH by adding acidic or alkaline solution in column 6, lines 33-37; organic reducing agent in column 8, lines 38-58. It would have been obvious to use reducing agent in the aqueous solution for reduction of light insensitive silver salt of an organic acid such as suggested therein to provide the claimed composition. The use of the basic solution such as ammonia to regulate the pH has been common in the art

Response to Arguments

6 Applicant's arguments filed December 9, 2002 have been fully considered but they are not persuasive because the reason set forth in the previous office action. It is also the Examiner's position that the additional argument submitted on June 2, 2002 is not persuasive. The step of forming the aqueous composition containing the organic silver salt and silver halide using pH higher than 8.0 is taught in Gilliam. There is no difference in processing step as how to increase the pH claimed in the present claimed invention and that taught in Gilliam. The scope of increasing the pH claimed in the claimed invention encompasses the step of adding the behenic acid with the solution of sodium hydroxide taught in Gilliam. The applicants fail to provide a convincing evidence that the pH of 8.7 taught in Gilliam decrease to less than pH 8.0 claimed in the present claimed invention. This argument with respect to pH of the prior art process is around 5.0 is based on the Cousels'assertion. The claims fails to state whether that the pH of at least 8.0 is maintaining troughtout the process. The specification on page 19, Table does not show that the prior the pH of the prior art is around 5.0 such as stated in the applicants'argument. Also, the argument with respect to step of increasing the pH

Art Unit: 1752

provided in the specification is not persuasive since the language or the processing presented in the specification is not read into the claimed invention. Although a claim should be interpreted in light of the specification disclosure, it is generally considered improper to read limitation contained in the specification into the claims. See In re Prater, 415 F.2d 1393, 162 USPQ 541 (CCPA 1969) and In re Winkhaus, 527 F.2d 637, 188 USPQ 129 (CCPA 1975), which discuss the premise that one cannot rely on the specification to impart limitation to the claim that are not recited in the claim. Counsel's arguments cannot take the place of evidence. In re Greenfield, 571 F. 2d 1185, 197 USPQ 227 (CCPA 1978).

The argument with the unexpected results is not persuasive since "(E)vidence of secondary considerations, such as unexpected results or commercial success, is irrelevant to 35 U.S.C 102 rejections and thus cannot overcome a rejection so based. In re Wiggins, 488 F.2d 538, 543, 179 USPQ 421, 425 (CCPA 1973). Note to claims 8, 10, 11, 15 wherein the silver halide is made by in-situe at pH of 8.7. Claims 5-6, 14 are related to the use of ex-situe photosensitive halide, the composition thereof is the same as taught in Gilliams even though they are made by different process. "(E)ven though product-by-process claims are limited by and defined by the process, determination of patentability is based on the product itself. The patentability of a product does not depend on its method of production. If the product in the product-by-process claim is the same or obvious from a product of prior art, the claim is unpatentable even though the prior art product was made by different process." In re Thorpe 777 F.2d 695, 698, 227 USPQ 694, 966 (Fed. Cir. 1985). The applicants fails to

Art Unit: 1752

show as to why the claimed material are patentably distinct over the material of Gilliams. The argument with respected to unexpected results is not persuasive the material presented in the specification was not prepared according to the teaching of Gilliams especially Exemple 1, and moreover, the argument with respect to the unexpected results is based on the Counsel'assertion. Counsel's arguments cannot take the place of evidence. In re Greenfield, 571 F. 2d 1185, 197 USPQ 227 (CCPA 1978).

7. Claims 1-4, 7, 17 are allowed over Gilliams since Gilliams fails to teach or fairly suggest the process of raising the pH of the aqueous dispersion to pH of at least 8.0 when the ex-situe photosensitive silver halide is used. The allowance of claims 8-10, and 13 set forth in the previous office action are withdrawn in view of newly found that Gilliams discloses to adjust pH of the aqueous dispersion to 8.7 such as set forth above.

Conclusion

8. **THIS ACTION IS MADE FINAL.** Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of

Application/Control Number: 09/934,806 Page 7

Art Unit: 1752

the advisory action. In no event, however, will the statutory period for reply expire later

than SIX MONTHS from the mailing date of this final action.

9. The prior art made of record and not relied upon is considered pertinent to

applicant's disclosure. The supplemental disclosure statement filed December 9, 2002

has been considered and made of record.

10. Any inquiry concerning this communication or earlier communications from the

examiner should be directed to Thorl Chea whose telephone number is (703)308-3498.

The examiner can normally be reached on M-F (9:30 - 6:00).

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Janet C Baxter can be reached on (703)308-2303. The fax phone numbers

tape mass, carrier or basis out to read the first (1.50) and the first out the mass of the first out the first out

for the organization where this application or proceeding is assigned are (703)872-9301

for regular communications and (703)872-9311 for After Final communications.

Any inquiry of a general nature or relating to the status of this application or

proceeding should be directed to the receptionist whose telephone number is (703)308-

0661.

tchea ()

August 11, 2003

Thorl Chea Primary Examiner

Art Unit 1752